PHYSICS

1 Polar molecules are the molecules:

1.	that acquires a dipole moment only when the magnetic field is absent.
2.	has a permanent electric dipole moment.
3.	has zero dipole moment.
4.	that acquire a dipole moment only in the presence of an electric field due to displacement of charges.

2 A particle is released from a height of S above the surface of the earth. At a certain height, its kinetic energy is three times its potential energy. The distance from the earth's surface and the speed of the particle at that instant are respectively:

1.	$\frac{S}{2}, \frac{\sqrt{3gS}}{2}$	2.	$\frac{S}{4}, \sqrt{\frac{3gS}{2}}$
3.	$\frac{S}{4}, \frac{3gS}{2}$	4.	$\frac{S}{4}, \frac{\sqrt{3gS}}{3}$

3 Match **Column I** and **Column II** with appropriate relations.

	Column I		**Column II**
(A)	Drift Velocity	(P)	$\frac{\mathrm{m}}{\mathrm{ne}^2\rho}$
(B)	Electrical Resistivity	(Q)	nev_d
(C)	Relaxation Period	(R)	$\frac{\mathrm{eE}}{\mathrm{m}}\tau$
(D)	Current Density	(S)	$\frac{E}{J}$

	(A)	**(B)**	**(C)**	**(D)**
1.	(R)	(P)	(S)	(Q)
2.	(R)	(Q)	(S)	(P)
3.	(R)	(S)	(P)	(Q)
4.	(R)	(S)	(Q)	(P)

4 If force $[F]$, acceleration $[A]$ and time $[T]$ are chosen as the fundamental physical quantities, then find the dimensions of energy:

1. $\left[FAT^{-1}\right]$
2. $\left[FA^{-1}T\right]$
3. $[FAT]$
4. $\left[FAT^2\right]$

5 The escape velocity from the Earth's surface is v. The escape velocity from the surface of another planet having a radius, four times that of Earth and the same mass density is:

1.	$3v$	2.	$4v$
3.	v	4.	$2v$

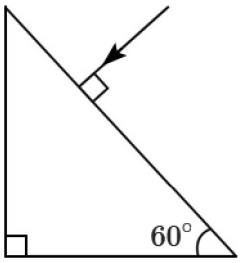
6 A small block slides down on a smooth inclined plane starting from rest at time $t = 0$. Let S_n be the distance traveled by the block in the interval $t = n - 1$ to $t = n$. Then the ratio $\frac{S_n}{S_{n+1}}$ is:

1.	$\frac{2n+1}{2n-1}$	2.	$\frac{2n}{2n-1}$
3.	$\frac{2n-1}{2n}$	4.	$\frac{2n-1}{2n+1}$

7 The velocity of a small ball of mass M and density d, when dropped in a container filled with glycerine becomes constant after some time. If the density of glycerine is $\frac{d}{2}$ then the viscous force acting on the ball will be:

1.	$\frac{3Mg}{2}$	2.	$2Mg$
3.	$\frac{Mg}{2}$	4.	Mg

8 Find the value of the angle of emergence from the prism given below for the incidence ray shown. The refractive index of the glass is $\sqrt{3}$.



1. 45°
2. 90°
3. 60°
4. 30°

9 A screw gauge gives the following readings when used to measure the diameter of a wire:
Main scale reading: 0 mm
Circular scale reading: 52 divisions
Given that 1 mm on the main scale corresponds to 100 divisions on the circular scale, the diameter of the wire that can be inferred from the given data is:

1.	0.26 cm	2.	0.052 cm
3.	0.52 cm	4.	0.026 cm

10 A nucleus with mass number 240 breaks into fragments each of mass number 120. The binding energy per nucleon of unfragmented nuclei is 7.6 MeV while that of fragments is 8.5 MeV. The total gain in the binding energy in the process is:
1. 804 MeV
2. 216 MeV
3. 0.9 MeV
4. 9.4 MeV

11 Water falls from a height of 60 m at the rate of 15 kg/s to operate a turbine. The losses due to frictional force are 10% of the input energy. How much power is generated by the turbine? $(g = 10)\ m/s^2$

1.	12.3 kW	2.	7.0 kW
3.	10.2 kW	4.	8.1 kW

12 A convex lens A of focal length 20 cm and a concave lens B of focal length 5 cm are kept along the same axis with a distance d between them. If a parallel beam of light falls on A leaves B as a parallel beam, then the distance d in (cm) will be:
1. 50
2. 30
3. 25
4. 15

13 The electron concentration in an n-type semiconductor is the same as the hole concentration in a p-type semiconductor. An external field (electric) is applied across each of them. Compare the currents in them.

1.	current in n-type $>$ current in p-type.
2.	no current will flow in p-type, current will only flow in n-type.
3.	current in n-type $=$ current in p-type.
4.	current in p-type $>$ current in n-type.

14 If a body is executing simple harmonic motion with frequency n, then the frequency of its potential energy is:

1.	$3n$	2.	$4n$
3.	n	4.	$2n$

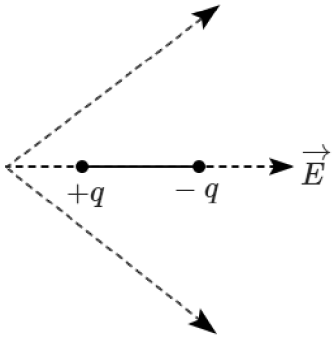
15 A cup of coffee cools from 90°C to 80°C in t minutes, when the room temperature is 20°C. The time taken by a similar cup of coffee to cool from 80°C to 60°C at room temperature same at 20°C is:
1. $\frac{10}{13}t$
2. $\frac{5}{13}t$
3. $\frac{13}{10}t$
4. $\frac{13}{5}t$

16 The effective resistance of a parallel connection that consists of four wires of equal length, equal area of cross-section, and same material is 0.25 Ω. What will be the effective resistance if they are connected in series?
1. 1 Ω
2. 4 Ω
3. 0.25 Ω
4. 0.5 Ω

17 A radioactive nucleus ${}^{A}_{Z}X$ undergoes spontaneous decay in the sequence ${}^{A}_{Z}X \rightarrow B_{Z-1} \rightarrow C_{Z-3} \rightarrow D_{Z-2}$ where Z is the atomic number of element X. The possible decay particles in the sequence are:

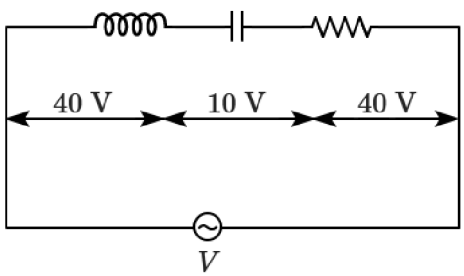
1.	$\beta^+,\ \alpha,\ \beta^-$	2.	$\beta^-,\ \alpha,\ \beta^+$
3.	$\alpha,\ \beta^-,\ \beta^+$	4.	$\alpha,\ \beta^+,\ \beta^-$

18 A dipole is placed in an electric field as shown. In which direction will it move?



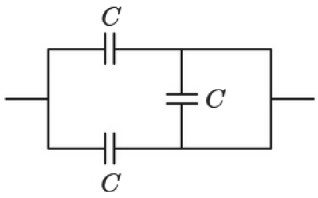
1.	towards the left as its potential energy will decrease.
2.	towards the right as its potential energy will increase.
3.	towards the left as its potential energy will increase.
4.	towards the right as its potential energy will decrease.

19 An inductor of inductance L, a capacitor of capacitance C and a resistor of resistance R are connected in series to an AC source of potential difference V volts as shown in Figure. The potential difference across L, C and R is 40 V, 10 V and 40 V, respectively. The amplitude of the current flowing through the LCR series circuit is $10\sqrt{2}$ A. The impedance of the circuit will be:



1.	$4\ \Omega$	2.	$5\ \Omega$
3.	$4\sqrt{2}\ \Omega$	4.	$\frac{5}{\sqrt{2}}\ \Omega$

20 The equivalent capacitance of the combination shown in the figure is:



1.	$\frac{C}{2}$	2.	$\frac{3C}{2}$
3.	$3C$	4.	$2C$

21 Given below are two statements:

Statement A:	A Zener diode is connected in reverse bias when used as a voltage regulator.
Statement B:	The potential barrier of p-n junction lies between 0.2 V to 0.3 V.

1.	**Statement A** is correct and **Statement B** is incorrect.
2.	**Statement A** is incorrect and **Statement B** is correct.
3.	**Statement A** and **Statement B** both are correct.
4.	**Statement A** and **Statement B** both are incorrect.

22 For a plane electromagnetic wave propagating in the x-direction, which one of the following combinations gives the correct possible directions for the electric field (E) and magnetic field (B) respectively?

1. $\hat{j}+\hat{k},\ -\hat{j}-\hat{k}$
2. $-\hat{j}+\hat{k},\ -\hat{j}+\hat{k}$
3. $\hat{j}+\hat{k},\ \hat{j}+\hat{k}$
4. $-\hat{j}+\hat{k},\ -\hat{j}-\hat{k}$

23 The number of photons per second on an average emitted by a source of monochromatic light of wavelength 600 nm, when it delivers the power of 3.3×10^{-3} watt will be:
$(h = 6.6 \times 10^{-34}$ J-s)

1.	10^{16}	2.	10^{15}
3.	10^{18}	4.	10^{17}

24 A parallel plate capacitor has a uniform electric field $\vec{E}$ in the space between the plates. If the distance between the plates is d and the area of each plate is A the energy stored in the capacitor is:
$(\varepsilon_0 =$ permittivity of free space$)$
1. $\frac{1}{2}\varepsilon_0 E^2 Ad$
2. $\frac{E^2 Ad}{\varepsilon_0}$
3. $\frac{1}{2}\varepsilon_0 E^2$
4. $\varepsilon_0 EAd$

25 A capacitor of capacitance C is connected across an AC source of voltage V, given by;
$V = V_0 \sin \omega t$
The displacement current between the plates of the capacitor would then be given by:
1. $I_d = \dfrac{V_0}{\omega C}\sin \omega t$
2. $I_d = V_0 \omega C \sin \omega t$
3. $I_d = V_0 \omega C \cos \omega t$
4. $I_d = \dfrac{V_0}{\omega C}\cos \omega t$

26 Two charged spherical conductors of radii R_1 and R_2 are connected by a wire. The ratio of surface charge densities of spheres $\left(\frac{\sigma_1}{\sigma_2}\right)$ is:

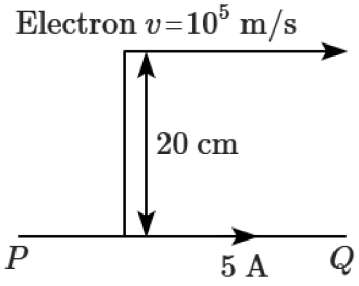
1.	$\sqrt{\dfrac{R_1}{R_2}}$	2.	$\dfrac{R_1^2}{R_2^2}$
3.	$\dfrac{R_1}{R_2}$	4.	$\dfrac{R_2}{R_1}$

27 An electromagnetic wave of wavelength λ is incident on a photosensitive surface of negligible work function. If 'm' is the mass of photoelectron emitted from the surface and λ_d is the de-Broglie wavelength, then:

1.	$\lambda = \left(\frac{2mc}{h}\right)\lambda_d^2$	2.	$\lambda = \left(\frac{2h}{mc}\right)\lambda_d^2$
3.	$\lambda = \left(\frac{2m}{hc}\right)\lambda_d^2$	4.	$\lambda_d = \left(\frac{2mc}{h}\right)\lambda^2$

28 A spring is stretched by 5 cm by a force 10 N. The time period of the oscillations when a mass of 2 kg is suspended by it is:
1. 3.14 s
2. 0.628 s
3. 0.0628 s
4. 6.28 s

29 An infinitely long straight conductor carries a current of 5 A as shown. An electron is moving with a speed of 10^5 m/s parallel to the conductor. The perpendicular distance between the electron and the conductor is 20 cm at an instant. Calculate the magnitude of the force experienced by the electron at that instant.



1. $4\pi \times 10^{-20}$ N
2. 8×10^{-20} N
3. 4×10^{-20} N
4. $8\pi \times 10^{-20}$ N

30 Match **Column I** and **Column II** and choose the correct match from the given choices.

	Column I		Column II
(A)	Root mean square speed of gas molecules	(P)	$\frac{1}{3}nm\bar{v}^2$
(B)	The pressure exerted by an ideal gas	(Q)	$\sqrt{\frac{3RT}{M}}$
(C)	The average kinetic energy of a molecule	(R)	$\frac{5}{2}RT$
(D)	The total internal energy of a mole of a diatomic gas	(S)	$\frac{3}{2}k_BT$

	(A)	(B)	(C)	(D)
1.	(Q)	(P)	(S)	(R)
2.	(R)	(Q)	(P)	(S)
3.	(R)	(P)	(S)	(Q)
4.	(Q)	(R)	(S)	(P)

31 If E and G respectively, denote energy and gravitational constant, then $\frac{E}{G}$ has the dimensions of:

1. $[ML^0T^0]$
2. $[M^2L^{-2}T^{-1}]$
3. $[M^2L^{-1}T^0]$
4. $[ML^{-1}T^{-1}]$

32 From a circular ring of mass M and radius R, an arc corresponding to a 90° sector is removed. The moment of inertia of the remaining part of the ring about an axis passing through the centre of the ring and perpendicular to the plane of the ring is K times MR^2. The value of K will be:

1.	$\frac{1}{4}$	2.	$\frac{1}{8}$
3.	$\frac{3}{4}$	4.	$\frac{7}{8}$

33 A uniform conducting wire of length $12a$ and resistance 'R' is wound up as a current-carrying coil in the shape of;

(i)	an equilateral triangle of side 'a'
(ii)	a square of side 'a'

The magnetic dipole moments of the coil in each case respectively are:

1. $3Ia^2$ and $4Ia^2$
2. $4Ia^2$ and $3Ia^2$
3. $\sqrt{3}Ia^2$ and $3Ia^2$
4. $3Ia^2$ and Ia^2

34 Two conducting circular loops of radii R_1 and R_2 are placed in the same plane with their centres coinciding. If $R_1 >> R_2$ the mutual inductance M between them will be directly proportional to:

1. $\frac{R_1^2}{R_2}$
2. $\frac{R_2^2}{R_1}$
3. $\frac{R_1}{R_2}$
4. $\frac{R_2}{R_1}$

35 A ball of mass 0.15 kg is dropped from a height 10 m, strikes the ground, and rebounds to the same height. The magnitude of impulse imparted to the ball is $(g = 10\ \text{m/s}^2)$ nearly:

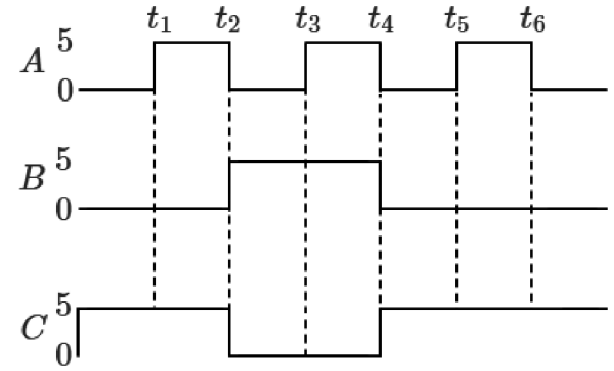
1. 2.1 kg-m/s
2. 1.4 kg-m/s
3. 0 kg-m/s
4. 4.2 kg-m/s

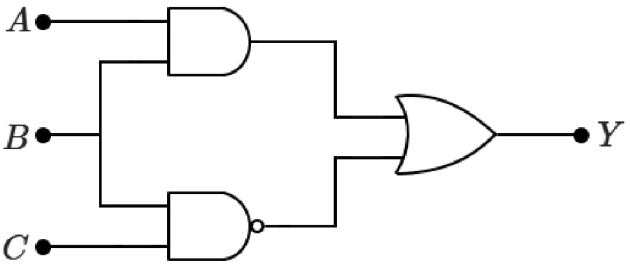
36 A step-down transformer connected to an AC mains supply of 220 V is made to operate at 11 V, 44 W lamp. Ignoring power losses in the transformer, what is the current in the primary circuit?

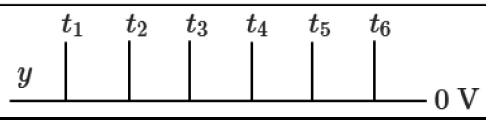
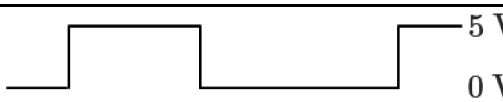
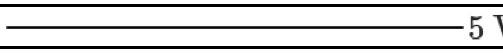
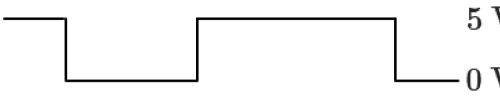
1. 2 A
2. 4 A
3. 0.2 A
4. 0.4 A

37 Twenty seven drops of same size are charged at 220 V each. They combine to form a bigger drop. Calculate the potential of the bigger drop:
1. 1520 V
2. 1980 V
3. 660 V
4. 1320 V

38 For the given circuit, the input digital signals are applied at the terminals A, B and C. What would be the output at terminal Y?





1.	
2.	
3.	
4.	

39 A particle of mass m is projected with a velocity, $v = kv_e\ (k < 1)$ from the surface of the earth. The maximum height, above the surface, reached by the particle is:
(Where $v_e =$ escape velocity, $R =$ the radius of the earth)

1.	$\dfrac{R^2k}{1+k}$	2.	$\dfrac{Rk^2}{1-k^2}$
3.	$R\left(\dfrac{k}{1-k}\right)^2$	4.	$R\left(\dfrac{k}{1+k}\right)^2$

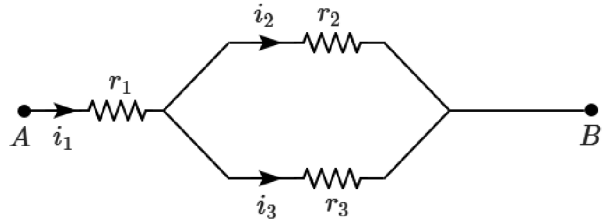
40 A series LCR circuit containing 5.0 H inductor, 80 μF capacitor and 40 Ω resistor is connected to 230 V variable frequency AC source. The angular frequencies of the source at which power transferred to the circuit is half the power at the resonant angular frequency are likely to be:

1.	46 rad/s and 54 rad/s
2.	42 rad/s and 58 rad/s
3.	25 rad/s and 75 rad/s
4.	50 rad/s and 25 rad/s

41 A particle moving in a circle of radius R with a uniform speed takes a time T to complete one revolution. If this particle were projected with the same speed at an angle θ to the horizontal, the maximum height attained by it equals $4R$. The angle of projection, θ is then given by:

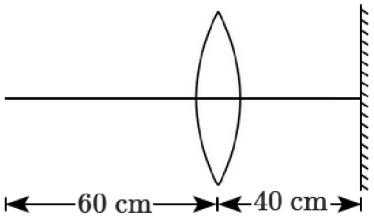
1.	$\theta = \sin^{-1}\left(\dfrac{\pi^2R}{gT^2}\right)^{1/2}$	2.	$\theta = \sin^{-1}\left(\dfrac{2gT^2}{\pi^2R}\right)^{1/2}$
3.	$\theta = \cos^{-1}\left(\dfrac{gT^2}{\pi^2R}\right)^{1/2}$	4.	$\theta = \cos^{-1}\left(\dfrac{\pi^2R}{gT^2}\right)^{1/2}$

42 Three resistors having resistances r_1, r_2 and r_3 are connected as shown in the given circuit. The ratio $\frac{i_3}{i_1}$ of currents in terms of resistances used in the circuit is:



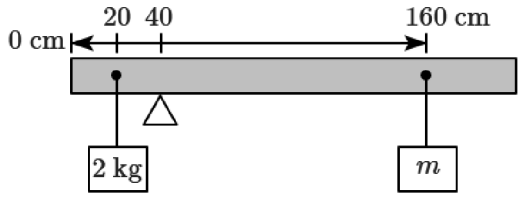
1. $\frac{r_1}{r_1+r_2}$
2. $\frac{r_2}{r_1+r_3}$
3. $\frac{r_1}{r_2+r_3}$
4. $\frac{r_2}{r_2+r_3}$

43 A point object is placed at a distance of 60 cm from a convex lens of focal length 30 cm. If a plane mirror were put perpendicular to the principal axis of the lens and at a distance of 40 cm from it, the final image would be formed at a distance of:



1.	30 cm from the plane mirror, it would be a virtual image.
2.	20 cm from the plane mirror, it would be a virtual image.
3.	20 cm from the lens, it would be a real image.
4.	30 cm from the lens, it would be a real image.

44 A uniform rod of length 200 cm and mass 500 g is balanced on a wedge placed at 40 cm mark. A mass of 2 kg is suspended from the rod at 20 cm and another unknown mass m is suspended from the rod at 160 cm mark as shown in the figure. What would be the value of m such that the rod is in equilibrium?
(Take $g = 10$ (m/s^2)



1.	$\frac{1}{6}$ kg	2.	$\frac{1}{12}$ kg
3.	$\frac{1}{2}$ kg	4.	$\frac{1}{3}$ kg

45 In the product

$$\vec{F} = q\left(\vec{v} \times \vec{B}\right)$$
$$= q\vec{v} \times \left(B\hat{i} + B\hat{j} + B_0\hat{k}\right)$$

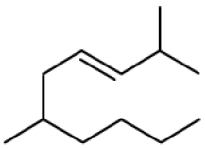
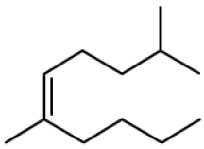
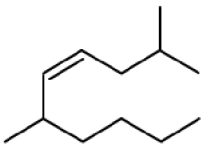
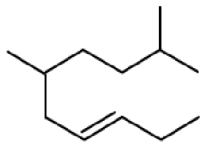
For $q = 1$ and $\vec{v} = 2\hat{i} + 4\hat{j} + 6\hat{k}$
and $\vec{F} = 4\hat{i} - 20\hat{j} + 12\hat{k}$
What will be the complete expression for $\vec{B}$?
1. $8\hat{i} + 8\hat{j} - 6\hat{k}$
2. $6\hat{i} + 6\hat{j} - 8\hat{k}$
3. $-8\hat{i} - 8\hat{j} - 6\hat{k}$
4. $-6\hat{i} - 6\hat{j} - 8\hat{k}$

CHEMISTRY

46 The compound which shows metamerism is:

1.	C_3H_6O	2.	$C_4H_{10}O$
3.	C_5H_{12}	4.	C_3H_8O

47 What is the correct structure of 2,6-Dimethyl-dec-4-ene?

1.		2.	
3.		4.	

48 Find the correct sequence of the bond enthalpy of the "C-X" bond among the following:

1. CH_3-F < CH_3-Cl > CH_3-Br > CH_3-I
2. CH_3-Cl > CH_3-F > CH_3-Br > CH_3-I
3. CH_3-F < CH_3-Cl < CH_3-Br < CH_3-I
4. CH_3-F > CH_3-Cl > CH_3-Br > CH_3-I

49 Zr (Z = 40) and Hf (Z = 72) have similar atomic and ionic radii because of:

1.	Lanthanoid contraction
2.	Having similar chemical properties
3.	Belonging to the same group
4.	Diagonal relationship

50 Based on the provided equation, what volume of oxygen gas (at STP) is required for the complete combustion of 39 grams of liquid benzene?

$$C_6H_{6(l)} + \frac{15}{2}O_{2(g)} \longrightarrow 6CO_{2(g)} + 3H_2O_{(l)}$$

1. 11.2 litre
2. 22.4 litre
3. 84 litre
4. 168 litre

51 BF_3 is a planar and an electron deficient compound. Hybridization and number of electrons around the central atom, respectively are:

1. sp^2 and 6
2. sp^2 and 8
3. sp^3 and 4
4. sp^3 and 6

52 The following solutions were prepared by dissolving 10 g of glucose ($C_6H_{12}O_6$) in 250 ml of water (P_1), 10 g of urea (CH_4N_2O) in 250 ml of water (P_2) and 10 g of sucrose ($C_{12}H_{22}O_{11}$) in 250 ml of water (P_3). The decreasing order of osmotic pressures of these solutions is:

1.	$P_2 > P_3 > P_1$	2.	$P_3 > P_1 > P_2$
3.	$P_2 > P_1 > P_3$	4.	$P_1 > P_2 > P_3$

53 Match Column I with Column II and mark the correct options:

	Column I (Compounds)		**Column II (Configurations)**
A.	$TiCl_4$	1.	$e^3t_2^3$
B.	FeO_4^{2-}	2.	$e^2t_2^0$
C.	$FeCl_4^{2-}$	3.	$e^2t_2^3$
D.	$MnCl_4^{2-}$	4.	$e^0t_2^0$

1. A(4), B(2), C(1), D(3)
2. A(4), B(3), C(2), D(1)
3. A(1), B(2), C(3), D(4)
4. A(2), B(4), C(3), D(1)

54 Ethylene diaminetetraacetate (EDTA) ion is:

1.	Bidentate ligand with two "N" donor atoms
2.	Tridentate ligand with three "N" donor atoms
3.	Hexadentate ligand with four "O" and two "N" donor atoms
4.	Unidentate ligand

55 The IUPAC name of the product that formed in the following chemical reaction is:

$$\text{Acetone} \xrightarrow[\text{(ii) } H_2O,\ H^+]{\text{(i) } C_2H_5MgBr,\ \text{dry Ether}} \text{Product}$$

1.	Pentan-3-ol	2.	2-Methyl butan-2-ol
3.	2-Methyl propan-2-ol	4.	Pentan-2-ol

56 The major product formed in the dehydrohalogenation reaction of 2-Bromo pentane is Pent-2-ene. This product formation is based on:

1. Hofmann Rule
2. Huckel's Rule
3. Saytzeff's Rule
4. Hund's Rule

57 The correct example of metal displacement reaction among the following is:

1. $Fe + 2HCl \rightarrow FeCl_2 + H_2 \uparrow$
2. $2\,Pb\,(NO_3)_2 \rightarrow 2PbO + 4\,NO_2 + O_2 \uparrow$
3. $2\,KClO_3 \xrightarrow{\Delta} 2KCl + 3O_2$
4. $Cr_2\,O_3 + 2Al \xrightarrow{\Delta} Al_2\,O_3 + 2\,Cr$

58 Which of the following options correctly represents the relationship between C_p and C_V for one mole of an ideal gas?

1.	$C_P = RC_V$	2.	$C_V = RC_P$
3.	$C_P + C_V = R$	4.	$C_P - C_V = R$

59 The RBC deficiency is a deficiency disease of:

1. Vitamin B_1
2. Vitamin B_2
3. Vitamin B_{12}
4. Vitamin B_6

60 Which of the following oxides is not expected to react with sodium hydroxide?

1. BaO
2. SO_3
3. Al_2O_3
4. SiO_2

61 An organic compound contains 80 % (by wt.) carbon and the remaining percentage of hydrogen. The empirical formula of this compound is:
[Atomic wt. of C is 12, H is 1]

1.	CH_3	2.	CH_4
3.	CH	4.	CH_2

62 The dihedral angle of the least stable conformer of ethane is:

1.	60°	2.	0°
3.	120°	4.	180°

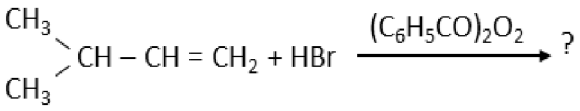
63 Match List - I with List - II

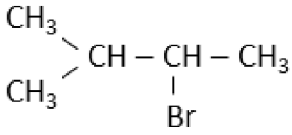
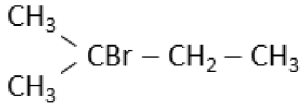
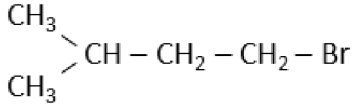
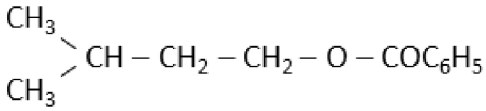
	List - I		**List - II**
(a)	PCl_5	(i)	Square pyramidal
(b)	SF_6	(ii)	Trigonal planar
(c)	BrF_5	(iii)	Octahedral
(d)	BF_3	(iv)	Trigonal bipyramidal

Choose the correct answer from the options given below:

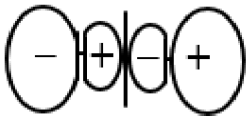
	(a)	**(b)**	**(c)**	**(d)**
1.	(iii)	(i)	(iv)	(ii)
2.	(iv)	(iii)	(ii)	(i)
3.	(iv)	(iii)	(i)	(ii)
4.	(ii)	(iii)	(iv)	(i)

64 The major product of the following chemical reaction is:



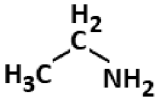
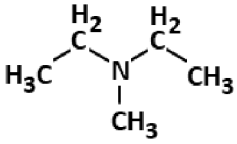
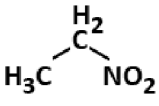
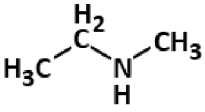
1.	
2.	
3.	
4.	

65 The molecular orbital diagram shown represents which of the following? (consider the z-axis as the internuclear axis)



1. $\sigma^{\star}2p_z$
2. $\pi 2p_{\mathrm{x}}$
3. $\pi^{\star}2p_{\mathrm{x}}$
4. $\sigma 2p_z$

66 Identify the compound that will react with Hinsberg's reagent to give a solid which dissolves in alkali.

1.		2.	
3.		4.	

67 In the light of the below statements, choose the correct answer from the options given below:

Statement I:	Acid strength increases in the order given as HF < HCl < HBr < HI.
Statement II:	As the size of the elements F, Cl, Br, I increases down the group, the bond strength of HF, HCI, HBr, and HI decreases and so the acid strength increases.

1.	**Statement I** is correct and **Statement II** is incorrect
2.	**Statement I** is incorrect and **Statement II** is correct.
3.	Both **Statement I** and **Statement II** are correct.
4.	Both **Statement I** and **Statement II** are incorrect.

68 The pK_b of dimethylamine and pk_a of acetic acid are 3.27 and 4.77 respectively at T (K).
The correct option for the pH of dimethylammonium acetate solution is:

1.	7.75	2.	6.25
3.	8.50	4.	5.50

69 The molar conductance of NaCl, HCI, and CH_3COONa at infinite dilution are 126.45, 426.16, and 91.0 S cm mol^{-1} respectively. The molar conductance of CH_3COOH at infinite dilution will be:

1. 698.28 S cm^2 mol^{-1}
2. 540.48 S cm^2 mol^{-1}
3. 201.28 S cm^2 mol^{-1}
4. 390.71 S cm^2 mol^{-1}

70 Given that the total pressure inside a reaction vessel is 1.12 atm at 105°C for the reaction $NH_4HS_{(s)} \rightleftharpoons NH_{3(g)} + H_2S_{(g)}$.
What is the value of K_p?

1. 0.56
2. 1.25
3. 0.31
4. 0.63

71 The vapour pressures of pure liquids A and B are 200 and 300 mmHg, respectively at 298K. On mixing the two liquids, the sum of their initial volumes is equal to the volume of the final mixture. The mole fraction of liquid B is 0.5 in the mixture. The vapour pressure of the final solution, the mole fractions of components A and B in vapour phase, respectively are:

1. 250 mmHg, 0.4, 0.6
2. 500 mmHg, 0.5, 0.5
3. 450 mmHg, 0.5, 0.5
4. 500 mmHg, 0.4, 06

72 What is the unit of molar conductivity?

1. $\Omega^{-1}\ cm^{-2}\ mol^{-1}$
2. $\Omega\ cm^{-2}\ mol^{-1}$
3. $\Omega^{-1}\ cm^{2}\ mol^{-1}$
4. $cm^2\ mol$

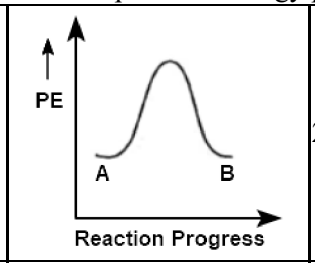
73 The incorrect statement among the following is:

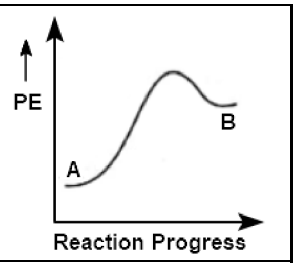
1.	Lanthanoids are good conductors of heat and electricity.
2.	Actinoids are highly reactive metals, especially when finely divided.
3.	Actinoid contraction is greater for element to element than Lanthanoid contraction.
4.	Most of the trivalent Lanthanoid ions are colorless in the solid-state.

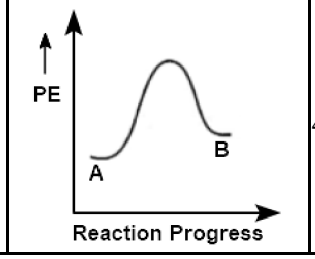
74 Choose the incorrect statement among the following regarding noble gases:

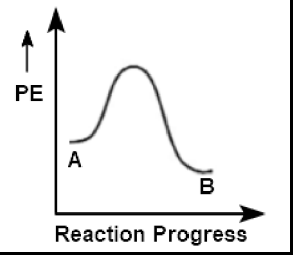
1.	Noble gases have weak dispersion forces.
2.	Noble gases have large positive values of electron gain enthalpy.
3.	Noble gases are sparingly soluble in water.
4.	Noble gases have very high melting and boiling points.

75 For a reaction A→B, enthalpy of reaction is -4.2 kJ mol^{-1} and enthalpy of activation is 9.6 kJ mol^{-1}. The correct potential energy profile for the reaction is:

1. 

2. 

3. 

4. 

76 Which lanthanoid symbol is incorrectly matched with its atomic number?
1. Cerium(Ce) (Z =58)
2. Promethium(Pm) (Z= 61)
3. Europium(Eu) (Z= 64)
4. Dysprosium(Dy) (Z = 66)

77 A particular station of All India Radio, New Delhi, broadcasts on a frequency of 1,368 kHz (kilohertz). The wavelength of the electromagnetic radiation emitted by the transmitter is: [speed of light, $c = 3.0 \times 10^8$ ms^{-1}]

1.	2192 m	2.	21.92 cm
3.	219.3 m	4.	219.2 m

78 In which one of the following arrangements the given sequence is not strictly according to the properties indicated against it?

1.	$NH_3 < PH_3 < AsH_3 < SbH_3$	Increasing acidic character
2.	$CO_2 < SiO_2 < SnO_2 < PbO_2$	Increasing oxidizing power
3.	HF < HCl < HBr < HI	Increasing acidic strength
4.	$H_2O < H_2S < H_2Se < H_2Te$	Increasing pK_a values

79 For irreversible expansion of an ideal gas under isothermal condition, the correct option is :
1. $\Delta U = 0,\ \Delta S_{total} \neq 0$
2. $\Delta U \neq 0,\ \Delta S_{total} = 0$
3. $\Delta U = 0,\ \Delta S_{total} = 0$
4. $\Delta U \neq 0,\ \Delta S_{total} \neq 0$

80 Which pair of ions from the following list do not constitute an iso-electronic pair?
1. Mn^{2+}, Fe^{3+}
2. Fe^{2+}, Mn^{2+}
3. O^{2-}, F^-
4. Na^+, Mg^{2+}

81 Consider the below reaction and identify the missing reagent/chemical.

$$CH_3CH_2COO^-Na^+ \xrightarrow[\text{Heat}]{\text{NaOH, + ?}} CH_3CH_3 + Na_2CO_3$$

1. CaO
2. DIBAL-H
3. B_2H_6
4. Red Phosphorus

82 Which molecule among the following is non-polar?

1.	$SbCl_5$	2.	NO_2
3.	$POCl_3$	4.	CH_2O

83 Match Column I with Column II, Columns I and II represent complexes and magnetic moment (BM) respectively.

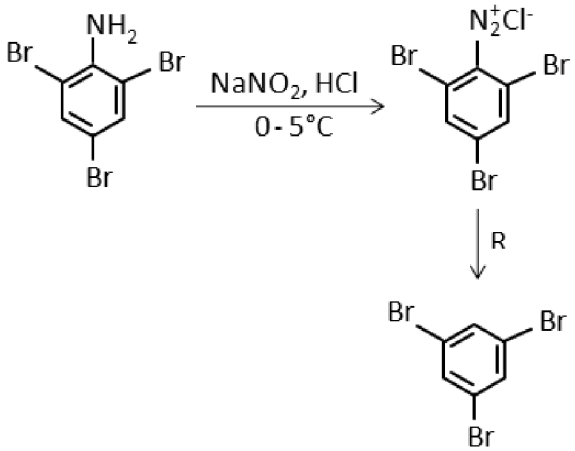
Column I	Column II
(a) $[FeCN_6]^{3-}$	(i) 5.92 BM
(b) $[Fe(H_2O)_6]^{3+}$	(ii) 0 BM
(c) $[Fe(CN)_6]^{4-}$	(iii) 4.90 BM
(d) $[Fe(H_2O)_6]^{2+}$	(iv) 1.73 BM

Choose the correct answer from the options given below:

	(a)	(b)	(c)	(d)
1.	(i)	(iii)	(iv)	(ii)
2.	(iv)	(i)	(ii)	(iii)
3.	(iv)	(ii)	(i)	(iii)
4.	(ii)	(iv)	(iii)	(i)

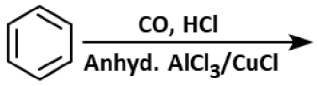
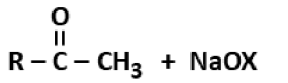
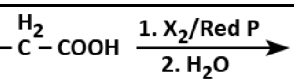
84 The correct option for the value of vapour pressure of a solution at 45 °C with benzene to octane in a molar ratio 3:2 is:
[At 45 °C vapour pressure of benzene is 280 mm Hg and that of octane is 420 mm Hg. Assume Ideal gas]
1. 336 mm of Hg
2. 350 mm of Hg
3. 160 mm of Hg
4. 168 mm of Hg

85 The reagent 'R' in the given sequence of a chemical reaction is :



1.	HI	2.	CuCN/KCN
3.	H_2O	4.	CH_3CH_2OH

86 Match List-I with List-II.

	List-I (Substrates of reactions)		List II (Name of reactions)
(a)		(i)	Hell-Volhard-Zelinsky reaction
(b)		(ii)	Gattermann-Koch reaction
(c)		(iii)	Haloform reaction
(d)		(iv)	Esterification

Choose the correct answer from the options given below:

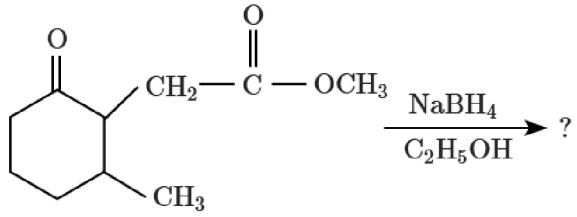
	(a)	(b)	(c)	(d)
1.	(i)	(iv)	(iii)	(ii)
2.	(ii)	(iii)	(iv)	(i)
3.	(iv)	(i)	(ii)	(iii)
4.	(iii)	(ii)	(i)	(iv)

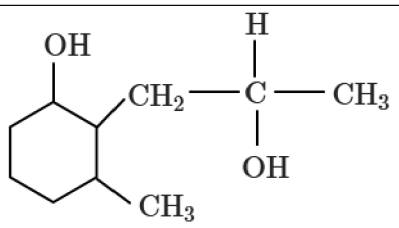
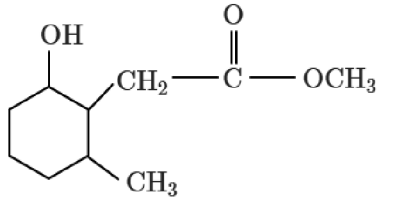
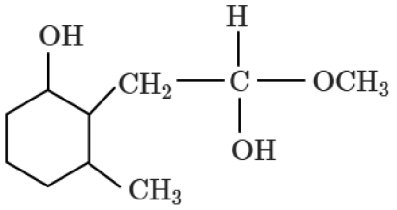
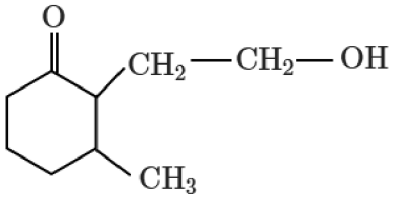
87 The molar conductivity of 0.007 M acetic acid is 20 S cm^2 mol^{-1}. The dissociation constant of acetic acid is :
($\Lambda^o_{H^+} = 350$ S cm^2 mol^{-1})
($\Lambda^o_{CH_3COO^-} = 50$ S cm^2 mol^{-1})
1. 1.75×10^{-5} mol L^{-1}
2. 2.50×10^{-5} mol L^{-1}
3. 1.75×10^{-4} mol L^{-1}
4. 2.50×10^{-4} mol L^{-1}

88 The slope of Arrhenius Plot (ln k v/s $\frac{1}{T}$) of the first-order reaction is -5×10^3 K. The value of E_a of the reaction is:
[Given R = 8.314 JK^{-1} mol^{-1}]

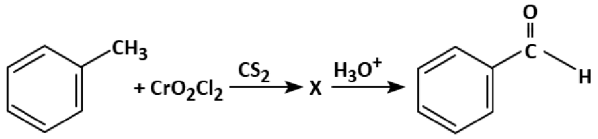
1.	166 kJ mol^{-1}	2.	–83 kJ mol^{-1}
3.	41.5 kJ mol^{-1}	4.	83.0 kJ mol^{-1}

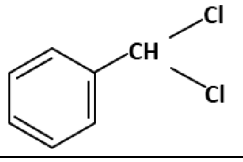
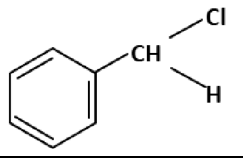
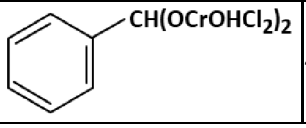
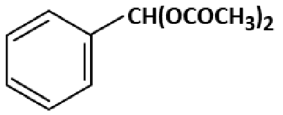
89 The product formed in the following chemical reaction is:



1.	
2.	
3.	
4.	

90 The intermediate compound 'X' in the following chemical reaction is :



1.		2.	
3.		4.	

BIOLOGY

91 When the centromere is situated in the middle of two equal arms of chromosomes, the chromosome is referred as :

1. Sub-metacentric
2. Acrocentric
3. Metacentric
4. Telocentric

92 In the equation GPP – R = NPP, R represents:

1. Environment factor
2. Respiration losses
3. Radiant energy
4. Retardation factor

93 Which of the following stages of meiosis involves division of centromere?

1. Anaphase II
2. Telophase II
3. Metaphase I
4. Metaphase II

94 Match List - I with List - II

	List -I		List - II
a.	Cristae	i.	Primary constriction in chromosome
b.	Thylakoids	ii.	Disc-shaped sacs in Golgi apparatus
c.	Centromere	iii.	Infoldings in mitochondria
d.	Cisternae	iv.	Flattened membranous sacs in stroma of plastids

Choose the correct answer from the options given below:

	(a)	(b)	(c)	(d)
1.	(iii)	(iv)	(i)	(ii)
2.	(ii)	(iii)	(iv)	(i)
3.	(iv)	(iii)	(ii)	(i)
4.	(i)	(iv)	(iii)	(ii)

95 Diadelphous stamens are found in:

1. Pea
2. China rose and citrus
3. China rose
4. *Citrus*

96 The term used for transfer of pollen grains from anthers of one plant to stigma of a different plant which during pollination, brings genetically different types of pollen grains to stigma, is:

1.	Chasmogamy	2.	Cleistogamy
3.	Xenogamy	4.	Geitonogamy

97 The plant hormone used to destroy weeds in a field is:

1.	2, 4-D	2.	IBA
3.	IAA	4.	NAA

98 Which of the following algae contains mannitol as reserve food material?

1.	*Volvox*	2.	*Ulothrix*
3.	*Ectocarpus*	4.	*Gracilaria*

99 DNA strands on a gel, stained with ethidium bromide, when viewed under UV radiation, appear as:

1.	Dark red bands	2.	Bright blue bands
3.	Yellow bands	4.	Bright orange bands

100 Which of the following are not secondary metabolites in plants?
1. Vinblastin, curcumin
2. Rubber, gums
3. Morphine, codeine
4. Amino acids, glucose

101 Inspite of interspecific competition in nature, which mechanism the competing species might have evolved for their survival?

1.	Mutualism	2.	Predation
3.	Resource partitioning	4.	Competitive release

102 Amensalism can be represented as:
1. Species A (–) : Species B (–)
2. Species A (+) : Species B (0)
3. Species A (–) : Species B (0)
4. Species A (+) : Species B (+)

103 A typical angiosperm embryo sac at maturity is:
1. 7-nucleate and 7-celled
2. 8-nucleate and 8-celled
3. 8-nucleate and 7-celled
4. 7-nucleate and 8-celled

104 The first stable product of CO_2 fixation in sorghum is:
1. Succinic acid
2. Phosphoglyceric acid
3. Pyruvic acid
4. Oxaloacetic acid

105 Plants follow different pathways in response to environment or phases of life to form different kinds of structures. This ability is called:

1.	Plasticity	2.	Maturity
3.	Elasticity	4.	Flexibility

106 Match List - I with List - II

List - I	**List - II**
(a) Protoplast fusion	(i) Totipotency
(b) Plant tissue culture	(ii) Pomato
(c) Meristem culture	(iii) Somaclones
(d) Micropropagation	(iv) Virus free plants

Choose the correct answer from the options given below:

	(a)	(b)	(c)	(d)
1.	(iii)	(iv)	(i)	(ii)
2.	(iv)	(iii)	(ii)	(i)
3.	(iii)	(iv)	(ii)	(i)
4.	(ii)	(i)	(iv)	(iii)

107 Which of the following is not an application of PCR (Polymerase Chain Reaction)?
1. Purification of isolated protein
2. Detection of gene mutation
3. Molecular diagnosis
4. Gene amplification

108 Which of the following algae produce Carrageen?

1.	Red algae	2.	Blue-green algae
3.	Green algae	4.	Brown algae

109 The production of gametes by the parents, formation of zygotes, the F_1 and F_2 plants, can be understood from a diagram called:
1. Punnett square
2. Net square
3. Bullet square
4. Punch square

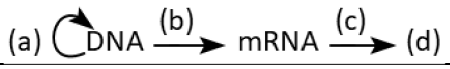
110 Which of the following is an incorrect statement?

1.	The perinuclear space forms a barrier between the materials present inside the nucleus and that of the cytoplasm.
2.	Nuclear pores act as passages for proteins and RNA molecules in both directions between nucleus and cytoplasm.
3.	Mature sieve tube elements possess a conspicuous nucleus and usual cytoplasmic organelles.
4.	Microbodies are present both in plant and animal cells.

111 Gemmae are present in:

1. Some Gymnosperms
2. Some Liverworts
3. Mosses
4. Pteridophytes

112 Complete the flow chart on central dogma



1.	(a) - Replication; (b) - Transcription; (c) - Translation; (d) - Protein
2.	(a) - Transduction; (b) - Translation; (c) - Replication; (d) - Protein
3.	(a) - Replication; (b) - Transcription; (c) - Transduction; (d) - Protein
4.	(a) - Transcription; (b) - Replication; (c) - Transcription; (d) - Transduction

113 Which of the following is a correct sequence of steps in a PCR (Polymerase Chain Reaction)?

1. Extension, Denaturation, Annealing
2. Annealing, Denaturation, Extension
3. Denaturation, Annealing, Extension
4. Denaturation, Extension, Annealing

114 During the purification process for recombinant DNA technology, addition of chilled ethanol precipitates out:

1.	Histones	2.	Polysaccharides
3.	RNA	4.	DNA

115 Which of the following statements is not correct?

1.	Pyramid of energy is always upright.
2.	Pyramid of numbers in a grassland ecosystem is upright.
3.	Pyramid of biomass in sea is generally inverted.
4.	Pyramid of biomass in sea is generally upright.

116 Match List-I with List-II.

	List-I		**List-I**
(a)	Lenticels	(i)	Phellogen
(b)	Cork cambium	(ii)	Suberin deposition
(c)	Secondary cortex	(iii)	Exchange of gases
(d)	Cork	(iv)	Phelloderm

Choose the correct answer from the options given below:

	(a)	(b)	(c)	(d)
1.	(ii)	(iii)	(iv)	(i)
2.	(iv)	(ii)	(i)	(iii)
3.	(iv)	(i)	(iii)	(ii)
4.	(iii)	(i)	(iv)	(ii)

117 Consider the given two statements:

Statement I:	Differentiation in plant cells is an irreversible process, under normal condition, where cells acquire specific structures to perform specialized functions.
Statement II:	Dedifferentiation in plants occurs when previously differentiated cells regain the capacity to divide under certain conditions, forming meristematic tissues like cork cambium.

Which of the following is correct?

1. Only **Statement I** is correct.
2. Only **Statement II** is correct.
3. Both **Statement I** and **Statement II** are correct.
4. Both **Statement I** and **Statement II** are incorrect.

118 The factor that leads to Founder effect in a population is:

1. Mutation
2. Genetic drift
3. Natural selection
4. Genetic recombination

119 In an ecosystem, during decomposition, bacterial and fungal enzymes degrade detritus into simpler inorganic substances. This process is called as:

1.	Fragmentation	2.	Leaching
3.	Catabolism	4.	Humification

120 When gene targeting involving gene amplification is attempted in an individual's tissue to treat disease, it is known as:

1.	Molecular diagnosis	2.	Safety testing
3.	Biopiracy	4.	Gene therapy

121 Genera like *Selaginella* and *Salvinia* produce two kinds of spores. Such plants are known as

1.	Homosporous	2.	Heterosporous
3.	Homosorus	4.	Heterosorus

122 What is the role of RNA polymerase III in the process of transcription in eukaryotes?

1. Transcribes precursor of mRNA
2. Transcribes only snRNAs
3. Transcribes rRNAs (28S, 18S and 5.8S)
4. Transcribes tRNA, 5s rRNA and snRNA

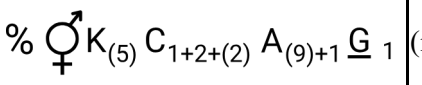
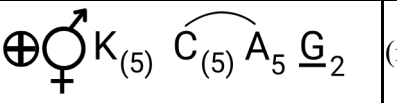
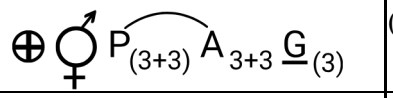
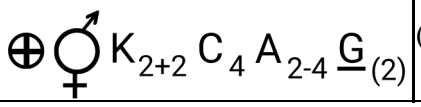
123 Which of the following statements is correct?

1.	Organisms that depend on living plants are called saprophytes.
2.	Some of the organisms can fix atmospheric nitrogen in specialized cells called sheath cells.
3.	The fusion of two cells is called Karyogamy.
4.	Fusion of protoplasms between two motile or non-motile gametes is called plasmogamy.

124 Plasmid pBR322 has a PstI restriction enzyme site within gene amp^R that confers ampicillin resistance. If this enzyme is used for inserting a gene for β-galactoside production and the recombinant plasmid is inserted in an E.coli strain,

1.	it will lead to the lysis of host cells.
2.	it will be able to produce a novel protein with dual abilities.
3.	it will not be able to confer ampicillin resistance to the host cell.
4.	the transformed cells will have the ability to resist ampicillin as well as produce β -galactoside.

125 Match Column I with Column II

	Column I		**Column II**
(a)		(i)	Brassicaceae
(b)		(ii)	Liliaceae
(c)		(iii)	Fabaceae
(d)		(iv)	Solanaceae

Select the correct answer from the options given below:

	(a)	**(b)**	**(c)**	**(d)**
1.	(ii)	(iii)	(iv)	(i)
2.	(iv)	(ii)	(i)	(iii)
3.	(iii)	(iv)	(ii)	(i)
4.	(i)	(ii)	(iii)	(iv)

126 Which of the following statements is incorrect?

1.	ATP is synthesized through complex V.
2.	Oxidation-reduction reactions produce a proton gradient in respiration.
3.	During aerobic respiration, the role of oxygen is limited to the terminal stage.
4.	In ETC (Electron Transport Chain), one molecule of $NADH+H^+$ gives rise to 2 ATP molecules, and one $FADH_2$ gives rise to 3 ATP molecules.

127 Match List -I with List - II.

	List-I		List-II
(a)	Protein	I.	C-C double bonds
(b)	Unsaturated fatty acid	II.	Phosphodiester bonds
(c)	Nucleic acid	III.	Glycosidic bonds
(d)	Polysaccharides	IV.	Peptide bonds

Choose the correct answer from the options given below:

Options:	(a)	(b)	(c)	(d)
1.	II	I	IV	III
2.	IV	III	I	II
3.	IV	I	II	III
4.	I	IV	III	II

128 DNA fingerprinting involves identifying differences in some specific regions in DNA sequence called:

1.	Single nucleotides	2.	Polymorphic DNA
3.	Satellite DNA	4.	Repetitive DNA

129 Match List-I with List-II

	List-I		List-II
(a)	S phase	i.	Proteins are synthesized
(b)	G_2 phase	ii.	Inactive phase
(c)	Quiescent stage	iii.	Interval between mitosis and initiation of DNA replication
(d)	G_1 phase	iv.	DNA replication

Choose the correct answer from the options given below:

Options:	(a)	(b)	(c)	(d)
1.	iv	i	ii	iii
2.	ii	iv	iii	i
3.	iii	ii	i	iv
4.	iv	ii	iii	i

130 Identify the correct statement:

1.	The coding strand in a transcription unit is copied to an mRNA.
2.	Split gene arrangement is characteristic of prokaryotes.
3.	In capping, methylguanosine triphosphate is added to the 3' end of hnRNA.
4.	RNA polymerase binds with the Rho factor to terminate the process of transcription in bacteria.

131 In roots

(a) Xylem is exarch type
(b) Protoxylem lies towards pith
(c) Xylem is endarch type
From the above statements, choose which one(s) is/are correct?
1. **a** and **b**
2. **b** and **c**
3. **a** only
4. **c** only

132 Which of the following statements is incorrect?

1.	Grana lamellae have both PS I and PS II.
2.	Cyclic photophosphorylation involves both PS I and PS II.
3.	Both ATP and NADPH + H^+ are non-cyclic synthesized during photophosphorylation.
4.	Stroma lamellae have PS I only and lack NADP reductase.

133 Nowadays, it is possible to detect the mutated gene causing cancer by allowing the radioactive probe to hybridise its complementary DNA in clone of cells, followed by its detection using autoradiography because:

1.	mutated gene does not appear on a photographic film as the probe has no complementarity with it.
2.	mutated gene does not appear on a photographic film as the probe has complementarity with it.
3.	mutated gene partially appears on a photographic film.
4.	mutated gene completely and clearly appears on a photographic film.

134 In the exponential growth equation $N_t = N_0 e^{rt}$, e represents :
1. The base of natural logarithms
2. The base of geometric logarithms
3. The base of number logarithms
4. The base of exponential logarithms

135 In some members of which of the following pairs of families, pollen grains retain their viability for months after release?
1. Poaceae; Solanaceae
2. Rosaceae; Leguminosae
3. Poaceae; Rosaceae
4. Poaceae; Leguminosae

136 Match List - I with List - II

	List - I		**List - II**
(a)	*Aspergillus* Niger	(i)	Acetic Acid
(b)	*Acetobacter aceti*	(ii)	Lactic Acid
(c)	*Clostridium bretylium*	(iii)	Citric Acid
(d)	*Lactobacillus*	(iv)	Butyric Acid

Choose the correct answer from the options given below:

	(a)	(b)	(c)	(d)
1.	(ii)	(iii)	(i)	(iv)
2.	(iv)	(ii)	(i)	(iii)
3.	(iii)	(i)	(iv)	(ii)
4.	(i)	(ii)	(iii)	(iv)

137 If Adenine makes 30% of the DNA molecule, what will be the percentage of Thymine, Guanine and Cytosine in it?

1.	T:30 ; G:20 ; C:20	2.	T:20 ; G:25 ; C:25
3.	T:20 ; G:30 ; C:20	4.	T:20 ; G:20 ; C:30

138 Which one of the following is an example of Hormone releasing IUD?

1.	Cu 7	2.	Multiload 375
3.	CuT	4.	LNG 20

139 Which one of the following characteristics is incorrect with respect to cockroaches?

1.	In females, 7^{th} - 9^{th} sterna together form a genital pouch.
2.	10^{th} abdominal segment in both sexes, bears a pair of anal cerci.
3.	A ring of gastric caeca is present at the junction of midgut and hind gut.
4.	Hypopharynx lies within the cavity enclosed by the mouth parts.

140 Identify the incorrect pair.

1. Lectins - Concanavalin A
2. Drugs - Ricin
3. Alkaloids - Codeine
4. Toxin - Abrin

141 Venereal diseases can spread through:

(a)	Using sterile needles
(b)	Transfusion of blood from an infected person
(c)	Infected mother to foetus
(d)	Kissing
(e)	Inheritance

Choose the correct answer from the options given below:

1.	**(b)** and **(c)** only	2.	**(a)** and **(c)** only
3.	**(a), (b)** and **(c)**	4.	**(b), (c)** and **(d)**

142 Chronic auto immune disorder affecting neuro muscular junction leading to fatigue, weakening and paralysis of skeletal muscle is called as:

1.	Myasthenia gravis	2.	Gout
3.	Arthritis	4.	Muscular dystrophy

143 With regard to insulin, choose the correct options:

(a)	C-peptide is not present in mature insulin.
(b)	The insulin produced by rDNA technology has C-peptide.
(c)	The pro-insulin has C-peptide.
(d)	A-peptide and B-peptide of insulin are interconnected by disulphide bridges.

Choose the correct answer from the options given below:

1.	**(a)**, **(c)** and **(d)** only	2.	**(a)** and **(d)** Only
3.	**(b)** and **(d)** only	4.	**(b)** and **(c)** only

144 Which of the following statements wrongly represents the nature of smooth muscle?

1.	Communication among the cells is performed by intercalated discs.
2.	These muscles are present in the wall of blood vessels.
3.	These muscles have no striations.
4.	They are involuntary muscles.

145 The centriole undergoes duplication during:

1.	Metaphase	2.	G_2 phase
3.	S-phase	4.	Prophase

146 The organelles that are included in the endomembrane system are:

1.	Golgi complex, Mitochondria, Ribosomes, and Lysosomes
2.	Golgi complex, Endoplasmic reticulum, Mitochondria, and Lysosomes
3.	The endoplasmic reticulum, Mitochondria, Ribosomes, and Lysosomes
4.	The endoplasmic reticulum, Golgi complex, Lysosomes, and Vacuoles

147 Which of the following RNAs is not required for the synthesis of protein?
1. rRNA
2. siRNA
3. mRNA
4. tRNA

148 Match List - I with List - II

List-I	**List-II**
(a) Metamerism	(i) Coelenterata
(b) Canal system	(ii) Ctenophora
(c) Comb Plates	(iii) Annelida
(d) Cnidoblasts	(iv) Porifera

Choose the correct answer from the options given below:

Options:	(a)	(b)	(c)	(d)
1.	iii	iv	ii	i
2.	iv	i	ii	iii
3.	iv	iii	i	ii
4.	iii	iv	i	ii

149 Which stage of meiotic prophase shows terminalisation of chiasmata as its distincitve feature?
1. Diakinesis
2. Pachytene
3. Leptotene
4. Zygotene

150 Persons with 'AB' blood group are called as "Universal recipients". This is due to:

1.	Presence of antibodies, anti-A and anti-B, on RBCs.
2.	Absence of antibodies, anti-A and anti-B, in plasma.
3.	Absence of antigens A and B on the surface of RBCs.
4.	Absence of antigens A and B in plasma.

151 The partial pressures (in mm Hg) of oxygen (O_2) and carbon dioxide (CO_2) at alveoli (the site of diffusion) are :
1. $pO_2 = 95$ and $pCO_2 = 40$
2. $pO_2 = 159$ and $pCO_2 = 0.3$
3. $pO_2 = 104$ and $pCO_2 = 40$
4. $pO_2 = 40$ and $pCO_2 = 45$

152 Receptors for sperm binding in mammals are present on:

1.	Perivitelline space	2.	Zona pellucida
3.	Corona Radiata	4.	Vitelline membrane

153 Match the following

	List-I		**List-II**
(a)	*Physalia*	i.	Pearl oyster
(b)	*Limulus*	ii.	Portuguese Man of War
(c)	*Ancylostoma*	iii.	Living fossil
(d)	*Pinctada*	iv.	Hookworm

Choose the correct answer from the options given below:

Options:	(a)	(b)	(c)	(d)
1.	ii	iii	iv	i
2.	i	iv	iii	ii
3.	ii	iii	i	iv
4.	iv	i	iii	ii

154 In a cross between a male and female, both heterozygous for sickle cell anaemia gene, what percentage of the progeny will be diseased?

1.	25%	2.	100%
3.	50%	4.	75%

155 Read the following statements.

(a)	Metagenesis is observed in Helminths.
(b)	Echinoderms are triploblastic and coelomate animals.
(c)	Round worms have organ-system level of body organization.
(d)	Comb plates present in ctenophores help in digestion.
(e)	Water vascular system is characteristic of Echinoderms.

Choose the correct answer from the options given below:
1. **(a)**, **(d)** and **(e)** are correct
2. **(b)**, **(c)** and **(e)** are correct
3. **(c)**, **(d)** and **(e)** are correct
4. **(a)**, **(b)** and **(c)** are correct

156 Which enzyme is responsible for the conversion of inactive fibrinogens to fibrins?
1. Epinephrine
2. Thrombokinase
3. Thrombin
4. Renin

157 Which one of the following belongs to the family Muscidae?

1.	Cockroach	2.	House fly
3.	Fire fly	4.	Grasshopper

158 The fruit fly has 8 chromosomes (2n) in each cell. During the interphase of mitosis, if the number of chromosomes at G_1 phase is 8, what would be the number of chromosomes after the S phase?

1.	4	2.	32
3.	8	4.	16

159 A specific recognition sequence identified by endonucleases to make cuts at specific positions within the DNA is:
1. Palindromic Nucleotide sequences
2. Poly(A) tail sequences
3. Degenerate primer sequence
4. Okazaki sequences

160 Match List-I with List-II

	List-I		**List-II**
(a)	Vaults	(i)	Entry of sperm through the Cervix is blocked
(b)	IUDs	(ii)	Removal of Vas deferens
(c)	Vasectomy	(iii)	Phagocytosis of sperms within the Uterus
(d)	Tubectomy	(iv)	Removal of the fallopian tube

Choose the correct answer from the options given below:

	(a)	(b)	(c)	(d)
1.	(ii)	(iv)	(iii)	(i)
2.	(iii)	(i)	(iv)	(ii)
3.	(iv)	(ii)	(i)	(iii)
4.	(i)	(iii)	(ii)	(iv)

161 For effective treatment of the disease, early diagnosis and understanding of its pathophysiology is very important. Which of the following molecular diagnostic techniques is very useful for early detection?
1. ELISA Technique
2. Hybridization Technique
3. Western Blotting Technique
4. Southern Blotting Technique

162 Select the favorable conditions required for the formation of oxyhemoglobin at the alveoli.
1. High pO_2, high pCO_2, less H^+, higher temperature
2. Low pO_2, low pCO_2, more H^+, higher temperature
3. High pO_2, low pCO_2, less H^+, lower temperature
4. Low pO_2 high pCO_2 more H^+, higher temperature

163 During the process of gene amplification using PCR, if a very high temperature is not maintained in the beginning, then which of the following PCR will be affected first?

1.	Denaturation	2.	Ligation
3.	Annealing	4.	Extension

164 Which is the "Only enzyme" that has the "Capability" to catalyze Initiation, Elongation, and Termination in the process of transcription in prokaryotes?
1. DNA Ligase
2. DNase
3. DNA-dependent DNA polymerase
4. DNA-dependent RNA polymerase

165 Which one of the following organisms bears hollow and pneumatic long bones?
1. *Macropus*
2. *Ornithorhynchus*
3. *Neophron*
4. *Hemidactylus*

166 Erythropoietin, hormone which stimulates R.B.C. formation, is produced by:

1.	The cells of bone marrow
2.	Juxtaglomerular cells of the kidney
3.	Alpha cells of the pancreas
4.	The cells of the rostral adenohypophysis

167 Which of the following secretes the hormone, relaxin, during the later phase of pregnancy?
1. Foetus
2. Uterus
3. Graafian follicle
4. Corpus luteum

168 The Adenosine deaminase deficiency results into:

1.	Digestive disorder	2.	Addison's disease
3.	Dysfunction of Immune system	4.	Parkinson's disease

169 Which of these is not an important component of initiation of parturition in humans?
1. Release of oxytocin
2. Release of prolactin
3. Increase in estrogen and progesterone ratio
4. Synthesis of prostaglandins

170 Which one of the following statements about Histones is wrong?

1.	Histones are rich in amino acids - Lysine and Arginine.
2.	Histones carry a positive charge in the side chain.
3.	Histones are organized to form a unit of 8 molecules
4.	Histones are organized to form a unit of 10 molecules

171

Statement I:	The codon AUG' codes for methionine and phenylalanine.
Statement II:	AAA' and 'AAG are both codons that code for the amino acid lysine.

In light of the above statements, choose the correct answer from the options given below:

1.	**Statement I** is correct but **Statement II** is false.
2.	**Statement I** is incorrect but **Statement II** is true.
3.	Both **Statement I** and **Statement II** are true.
4.	Both **Statement I** and **Statement II** are false.

172 Following are the statements with reference to 'lipids'.

(a)	Lipids having only single bonds are called unsaturated fatty acids.
(b)	Lecithin is a phospholipid.
(c)	Trihydroxy propane is glycerol.
(d)	Palmitic acid has 20 carbon atoms including carboxyl carbon.
(e)	Arachidonic acid has 16 carbon atoms.

You have to choose the correct statements out of the five given above. Choose the correct answer from the options given below:
1. (b) and (c) only
2. (b) and (e) only
3. (a) and (b) only
4. (c) and (d) only

173 Following are the statements about prostomium of earthworm.

(a)	It serves as a covering for mouth.
(b)	It helps to open cracks in the soil into which it can crawl.
(c)	It is one of the sensory structures.
(d)	It is the first body segment.

Choose the correct answer from the options given below:
1. (a), (b), (c) and (d) are correct
2. (b) and (c) are correct
3. (a), (b) and (c) are correct
4. (a), (b) and (d) are correct

174 Consider the following statements about population density and select the correct answer:

1.	Population density can be measured through various methods, including biomass and percent cover.
2.	In large populations, indirect methods are often more practical than direct counts.
3.	For organisms like bacteria in a culture plate, counting individual cells is usually the preferred method of density measurement.

Choose the correct option:
1. Only **Statement 1** and **Statement 2** are correct.
2. Only **Statement 1** is correct.
3. Only **Statement 2** and **Statement 3** are correct.
4. All statements are correct.

175 Match List-I with List-II

	List-I		**List-II**
(a)	Scapula	(i)	Cartilaginous joints
(b)	Cranium	(ii)	Flat bone
(c)	Sternum	(iii)	Fibrous joints
(d)	Vertebral column	(iv)	Triangular flat bone

Choose the correct answer from the options given below:

	(a)	**(b)**	**(c)**	**(d)**
1.	(iv)	(ii)	(iii)	(i)
2.	(iv)	(iii)	(ii)	(i)
3.	(i)	(iii)	(ii)	(iv)
4.	(ii)	(iii)	(iv)	(i)

176 The population density of Siberian cranes at Bharatpur wetlands in a given year could be described as:

1.	Low, because their numbers are typically fewer than 10.
2.	High, because they congregate in large flocks.
3.	Average, as they share resources with other wetland species.
4.	Increasing, due to favourable habitat conditions.

177 Match List-I with List-II

	List-I		**List-II**
(a)	Filariasis	(i)	*Haemophilus influenzae*
(b)	Amoebiasis	(ii)	*Trichophyton*
(c)	Pneumonia	(iii)	*Wuchereria bancrofti*
(d)	Ringworm	(iv)	*Entamoeba histolytica*

Choose the correct answer from the options given below:

	(a)	**(b)**	**(c)**	**(d)**
1.	(i)	(ii)	(iv)	(iii)
2.	(ii)	(iii)	(i)	(iv)
3.	(iv)	(i)	(iii)	(ii)
4.	(iii)	(iv)	(i)	(ii)

178 Identify the types of cell junctions that help to stop the leakage of the substances across a tissue and facilitation of communication with neighbouring cells via rapid transfer of ions and molecules.
1. Adhering junctions and Tight junctions, respectively.
2. Adhering junctions and Gap junctions, respectively.
3. Gap junctions and Adhering junctions, respectively.
4. Tight junctions and Gap junctions, respectively.

179 During muscular contraction, which of the following events occur?

(a)	'H' zone disappears
(b)	'A' band widens
(c)	'I' band reduces in width
(d)	Myosin hydrolyzes ATP, releasing the ADP and Pi
(e)	Z-lines attached to actins are pulled inwards

Choose the correct answer from the options given below:
1. **(b), (c), (d), (e)** only
2. **(b), (d), (e), (a)** only
3. **(a), (c), (d), (e)** only
4. **(a), (b), (c), (d)** only

180 Match List-I with List-II:

	List-I		List-II
(a)	Adaptive radiation	(i)	Selection of resistant varieties due to excessive use of herbicides and pesticides
(b)	Convergent evolution	(ii)	Bones of forelimbs in Man and Whale
(c)	Divergent evolution	(iii)	Wings of Butterfly and Bird
(d)	Evolution by anthropogenic action	(iv)	Darwin Finches

Choose the correct answer from the options given below:

	(a)	(b)	(c)	(d)
1.	(ii)	(i)	(iv)	(iii)
2.	(i)	(iv)	(iii)	(ii)
3.	(iv)	(iii)	(ii)	(i)
4.	(iii)	(ii)	(i)	(iv)